9th Class 2017					
Chemistry		Group-II		Paper	
Time:	15 Minutes	(Objecti	ve Type)	Max. Marks: 42	
Note:	e: 15 Minutes (Objective Type) Max. Marks: 12 e: Four possible answers A, B, C and D to each question are given. The choice which you think is correct, fill that circle in front of that question with Marker or Pen ink in the answer-book. Cutting or filling two or more circles will result in zero mark in that question. Which one of the following solutions contain more				
1-1-	water:				
	(a) 2 M	(b)	1 M		
	(c) 0.5 M	(d)	0.25 M √		
2-	is a non-electrolyte:				
	(a) NaCl soluti				
	(c) Sugar solu	tion √ (d)	H ₂ SO ₄ sol	lution	
3-	The molecule which has triple covalent bond:				
	(a) H ₂		0,		
	(c) N ₂ 1/	(d)	C ₂ H ₄		
4-	The oxidation number of Cl in KClO ₃ is:			O. is	
	(a) +4		+5 √	3 10.	
	(c) +6		+7		
5-	The percentag		•	S:	
	(a) 70 %		.78 % √		
	(c) 21 %		20 %		
6- Mist is an example of solution:					
	(a) Liquid in ga	as √ (b)		id	
	(c) Solid in ga	s (d)	Gas in soli	d .	
7-	The mass of one molecule of water is:				
	(a) 18 amu √	(b)	19 amu		
	(c) 20 amu	(d)	25 amu		

8-	The gas with maxim	um rate of diffusion is:			
		(b) Hydrogen √			
	(c) Fluorine	(d) Chlorine			
9-	Who discovered atomic number:				
8	(a) Dalton	(b) Moseley √			
	(c) Rutherford	(d) Bohr			
10-	The isotope used to generate electricity in nuclear reactor is:				
	(a) C - 12	(b) U - 235 √			
	(c) Co - 60	(d) P - 32			
11-	The most reactive m				
		(b) Zinc			
	(c) Copper	(d) Gold			
12-	The shortest period of periodic table is:				
1 62 11	(a) Fourth	(b) Third			
	(c) Second	(d) First 1			